

Reg No.: _____

Name: _____

APJ ABDUL KALAM TECHNOLOGICAL UNIVERSITY
FOURTH SEMESTER B.TECH DEGREE EXAMINATION(R&S), MAY 2019

Course Code: FT208

**Course Name: ENGINEERING THERMODYNAMICS AND REACTION KINETICS
(FT)**

Max. Marks: 100

Duration: 3 Hours

PART A

Answer any threefull questions, each question carries 10 marks.

Marks

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|---|---|------|
| 1 | a) Define enthalpy. Derive the general expression for enthalpy. | (5) |
| | b) What is meant by a heat engine. | (5) |
| 2 | State the second law of thermodynamics. What are the limitations of first law of thermodynamics? | (10) |
| 3 | a) What is a system? Discuss about closed system, open system and isolated system with example. | (5) |
| | b) Explain the equivalence of Kelvin Plank's and Clausius statements of second law of thermodynamics. | (5) |
| 4 | a) Explain the P-V-T behaviour of pure fluids. | (5) |
| | b) From a reservoir at 600K, 1000 J of heat is transferred to an engine that operates on a carnot's cycle. The engine rejects heat to a reservoir at 300K. Determine the thermal efficiency of the cycle and the work done by the engine. | (5) |
| 5 | Explain the dependency of temperature on rate of reaction from collision theory. | (10) |
| 6 | a) Derive Maxwell's equations. | (10) |
| 7 | a) A rocket engine burns a stoichiometric mixture of fuel (liquid hydrogen) in oxidant (liquid oxygen) .The combustion chamber is cylindrical ,75 cm long and 60cm in diameter and the combustion produces 108 kg/s of exhaust gases. If combustion is complete find the rate of reaction hydrogen and of oxygen. | (10) |
| 8 | a) What is the activity of pure fluids? Explain the effect of pressure and temperature on activity of fluids. | (10) |

- 9 Explain the rate equation for irreversible second order reaction. (10)
- 10 Derive the equation for the time required for the conversion in ideal batch reactor (10)
- 11 With neat sketch explain the working of a continuous stirred tank reactor. (10)
- 12 Differentiate integral method of analysis and differential method of analysis in (10)
determine the rate of reactions.
- 13 Discuss the effect of different variables in enzymatic activity (10)
- 14 Discuss the Monod Equation in study of cell growth kinetics (10)
