

Reg No.: \_\_\_\_\_

Name: \_\_\_\_\_

**APJ ABDUL KALAM TECHNOLOGICAL UNIVERSITY**  
FIRST SEMESTER B.TECH DEGREE EXAMINATION, DECEMBER 2018

**Course Code: BE101-06**

**Course Name: INTRODUCTION TO CHEMICAL ENGINEERING**

Max. Marks: 100

Duration: 3 Hours

**PART A**

*Answer all questions, each carries 3 marks.*

Marks

- |   |  |     |
|---|--|-----|
| 1 | List out any three achievements of chemical engineers in fuelling the world's economy                                  | (3) |
| 2 | The superficial mass velocity is found to be 200 lb/ ft <sup>2</sup> h. Specify its equivalent in kg/ m <sup>2</sup> s | (3) |
| 3 | Distinguish between distillation and evaporation   | (3) |
| 4 | Explain different modes of heat transfer   | (3) |
| 5 | Enumerate the need for using U-tube manometer  | (3) |
| 6 | List any three flow measuring instruments  | (3) |
| 7 | List any six air pollutants.   | (3) |
| 8 | Give any three physical characteristics of wastewater.   | (3) |

**PART B**

*Answer eight questions, (at least one full question from each module) each carries 5 marks.*

**Module I**

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|----|---|-----|
| 9  | Classify chemical industries to give any five classes with an example   | (5) |
| 10 | Distinguish between batch and continuous operation. List the advantages of continuous operation over batch operation. | (5) |

**Module II**

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|----|--|-----|
| 11 | The heat transfer coefficient of oil flowing through a pipe is 300W/(.m <sup>2</sup> .K). Determine the value of heat transfer coefficient expressed in kcal/(h.m <sup>2</sup> °C), Btu/(h. ft <sup>2</sup> °F).   | (5) |
| 12 | One mole of a gas is contained in a cylinder of volume 0.5 m <sup>3</sup> at 200 kPa. Using van der Waals equation determines the temperature of the gas in cylinder. The van der Waals constants are $a = 0.135 \text{ Nm}^4/\text{mol}^2$ ; $b = 3.22 \times 10^{-5} \text{ m}^3/\text{mol}$ | (5) |

**Module III**

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| 13 | Explain saponification process and its industrial application | (5) |
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**Module IV**

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| 14 | Differentiate between Mixed flow reactor and Plug flow reactor. | (5) |
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**Module V**

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| 15 | With a schematic diagram explain working of a venturimeter | (5) |
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- 16 Explain DCDA process for the manufacturing of sulphuric acid with a process flow diagram (5)

**Module VI**

- 17 Discuss various types of solid waste management system (5)
- 18 List any five novel materials along with their application (5)

**PART C**

*Answer six questions, (at least one full question from each module) each carries 6 marks.*

**Module I**

- 19 Write the history and evolution of chemical engineering as a profession (6)

**Module II**

- 20 In the SI system, thermal conductivity has the unit W/(m K). The thermal conductivity of the solid material can be calculated as  $k = xQ/(A\Delta T)$ , where  $Q$  is the rate of heat transfer,  $x$  is the thickness of the solid,  $A$  is the area of heat transfer and  $\Delta T$  is the temperature difference across the solid. The following values were obtained experimentally:  $Q = 15000$  kJ/h,  $A = 50$  ft<sup>2</sup>,  $x = 100$  mm and  $\Delta T = 1000$  K (6)
- a) Calculate the thermal conductivity in W/(m K).
- b) Express the thermal conductivity in kcal/(h m °C)

**Module III**

- 21 a) Distinguish between unit operations and unit processes with an example. (3)
- b) Explain the principle of distillation. List any two types of distillation. (3)
- 22 Biodiesel can be used as an alternative to petroleum diesel. List out the various steps involved in the production of biodiesel (6)

**Module IV**

- 23 Identify the modes of heat transfer involved in the deep frying of food items in a frying pan. Justify your answer. (6)

**Module V**

- 24 a) Describe the basic concepts of a P&I diagram. (4)
- b) Draw the symbol of the following chemical apparatus and equipment used in Piping and Instrumentation diagram (i) reactor ii) heat exchanger (2)
- 25 Describe the principle and working of thermocouple. (6)

**Module VI**

- 26 Discuss the reasons lead to Bhopal gas tragedy (6)

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