Course code	Course Name	L-T-P- Credits	Year of Introduction
CH484	FUEL CELL TECHNOLOGY	3-0-0-3	2016

# Prerequisite : Nil Course Objectives

• To expose the students to the fundamental knowledge required in the development of fuel cell technology.

### **Syllabus**

Introduction to Fuel Cells and Fuel Cell Technology, General Thermodynamics, Reaction Kinetics, Charge and Mass Transport, Overview of Fuel Cell Types, Stack Design, Fuel Cell Characterization, Hydrogen Economy.

#### **Expected Outcome**

At the end of the course the students will be able to:

- 1. Know the fundamentals of electrochemistry, thermodynamics, fluid mechanics, and heat and mass transfer, appropriate for the design or review of components of fuel cells and fuel cell systems.
- 2. Analyze the fuel cell technology and compare different types of fuel cell systems.
- 3. Calculate the various losses in fuel cells and analyze the fuel cell power plant subsystems.
- 4. Defend the significance of fuel cell technology in the new global energy scenario.
- 5. Distinguish the expectances of hydrogen as a fuel and energy vector in the context of renewable energy.

#### **References Books:**

- 1. Andreas Zuttel; Andreas Borgschulte; Louis Schdaptach, Hydrogen as a future energy carrier, Wiley-VCH Verlag GmbH & Co., KGaA, Weinheim, 2008.
- 2. Costamagna, P.; Srinivasan, S, J Power Sources 2001, 102, 242-269...
- 3. Frano Barbir. PEM Fuel Cells: Theory and Practice. Elsevier, 2005
- 4. Fuel Cell Handbook,7the Edn., EG & G Technical Services, Nov 2004
- 5. Hordeski, M. F. Alternative Fuels: The Future of Hydrogen, The Fairmont Press: Lilburn, GA, 2007.
- 6. Kordesch, K.; Simader, G. Fuel Cells and Their Applications. VCH: 1996
- 7. Larminie, J.; Dicks, A. Fuel Cell Systems Explained. John Wiely & Sons Ltd: Chichester, 1999.
- 8. Ryan P. O'Hayre, Suk-Won Cha, Whitney Colella & Fritz B. Printz, Fuel Cell Fundamentals, John Wiley & Sons, Inc., New Jersey, 2006
- 9. Vielstich, W, Gasteiger, H. A. Lamm, A. (Eds):Handbook of Fuel Cells-Fundamentals, Technology and Applications. John Wiely & Sons Ltd: NY, 2003; Vols1-4

Course Plan					
Module	Contents	Hours	Sem. exam marks		
I	<b>Introduction:</b> Fuel Cell, Brief History of fuel cells, Types of Fuel Cells, Working of a PEM fuel Cell, Fuel Cell and conventional processes – comparison, Energy & power relations, units, Application scenarios, Advantages and disadvantages.	7	15%		

	General Thermodynamics: Enthalpy-Heat potential of fuel, Gibb's free energy-Work potential of fuel,		
	Reversible voltage - NERNST Equation, Voltage and P, T and concentration dependence – examples, Faraday's		
	Laws, Efficiency: thermodynamic, voltage and fuel.		
	Reaction Kinetics: Electrochemical reaction		
	fundamentals, electrode kinetics, Charge transfer and		
II	activations energy, Exchange current density - slow and		
	fast reactions, Potential and equilibrium - galvanic	N-7.4	150/
	potential, Reaction rate and potential - Butler Volmer	7	15%
	equation & Tafel equation, Electrocatalysts and reaction	AT	
	kinetics – typical exchange current densities, Electrode	AL	
	design basics	h. And	
	FIRST INTERNAL EXAMINATION		
	Charge and Mass Transport: Charge transport		
	resistances, voltage losses, Ionic and electronic		
	conductivites, Ionic conduction in different FC		
III	electrolytes: Aquesous, polymeric and ceramic, Diffusive	7	20%
	transport & voltage loss: Limiting current density,		
	Nerstian and kenetic effect, Convective transport: flow channels, gas diffusion / porous layer, gas velocity,		
	pressure, Flow channel configurations		
	Overview of Fuel Cell Types: PAFC, PEMFC, AFC,		
	MCFC, SOFC. Major Cell Components, Material	_	• • • •
IV	Properties, Processes and Operating Conditions of	7	20%
	PEMFC.		
	SECOND INTERNAL EXAMINATION		
	Stack Design: Sizing of a Fuel Cell Stack, Stack		
	Configuration, Uniform distribution of Reactants, Heat		
V	removal, Stack Clamping	7	15%
	Fuel Cell Diagnostics: Polarization Curve, Current	7	1070
	Interrupt, AC Impedance Spectroscopy, Pressure drop as		
	a diagnostic tool.		
	Fuel Cell System Design: Hydrogen-Oxygen Systems,	/	
	Hydrogen-Air Systems, Fuel Cell Systems with Fuel Processor, System Efficiency		
VI	Fuel Cells and Hydrogen Economy: Hydrogen Energy	7	15%
	Systems, Hydrogen Energy Technologies, Transition to		
	Hydrogen Economy		
	END SEMESTER EXAMINATION		

## **Question Paper Pattern**

Maximum Marks: 100 Exam Duration: 3 Hours

**Part A**: There shall be **Three questions** uniformly covering Modules 1 and 2, each carrying 15 marks, of which the student has to answer any **Two questions**. At the most 4 subdivisions can be there in one main question with a total of 15 marks for all the subdivisions put together.

 $(2 \times 15 = 30 \text{ Marks})$ 

**Part B**: There shall be **Three questions** uniformly covering Modules 3 and 4, each carrying 20 marks, of which the student has to answer any **Two questions**. At the most 4 subdivisions can be there in one main question with a total of 20 marks for all the subdivisions put together.

 $(2 \times 20 = 40 \text{ Marks})$ 

**Part C**: There shall be **Three questions** uniformly covering Modules 5 and 6, each carrying 15 marks, of which the student has to answer any **Two questions**. At the most 4 subdivisions can be there in one main question with a total of 15 marks for all the subdivisions put together.

 $(2 \times 15 = 30 \text{ Marks})$ 

